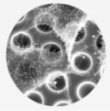


Protein adsorption in pores of ultrafiltration membranes

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Introduction

Membrane filtration

Pharmaceutical, food, petroleum, paper, ... industries

Separation process widely used for :

- desalting, purifying, decontaminating, concentrating, ...
- sea water, wastewaters, liquid food, ...

Advantages :

- easy control, energy efficient process, environmental respect

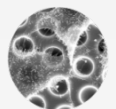


Introduction

Aim of this work

Studying adsorption phenomena in UF membrane (commercial TiO_2) and consequences on the process performances

- ✓ hydraulic performances
(filtrated flow rate)
- ✓ membrane selectivity
(separation effectiveness)



Introduction

Context

Solutions filtrated (sea water, wastewater, liquid food, ...)
Contain adhesive or viscous products as proteins

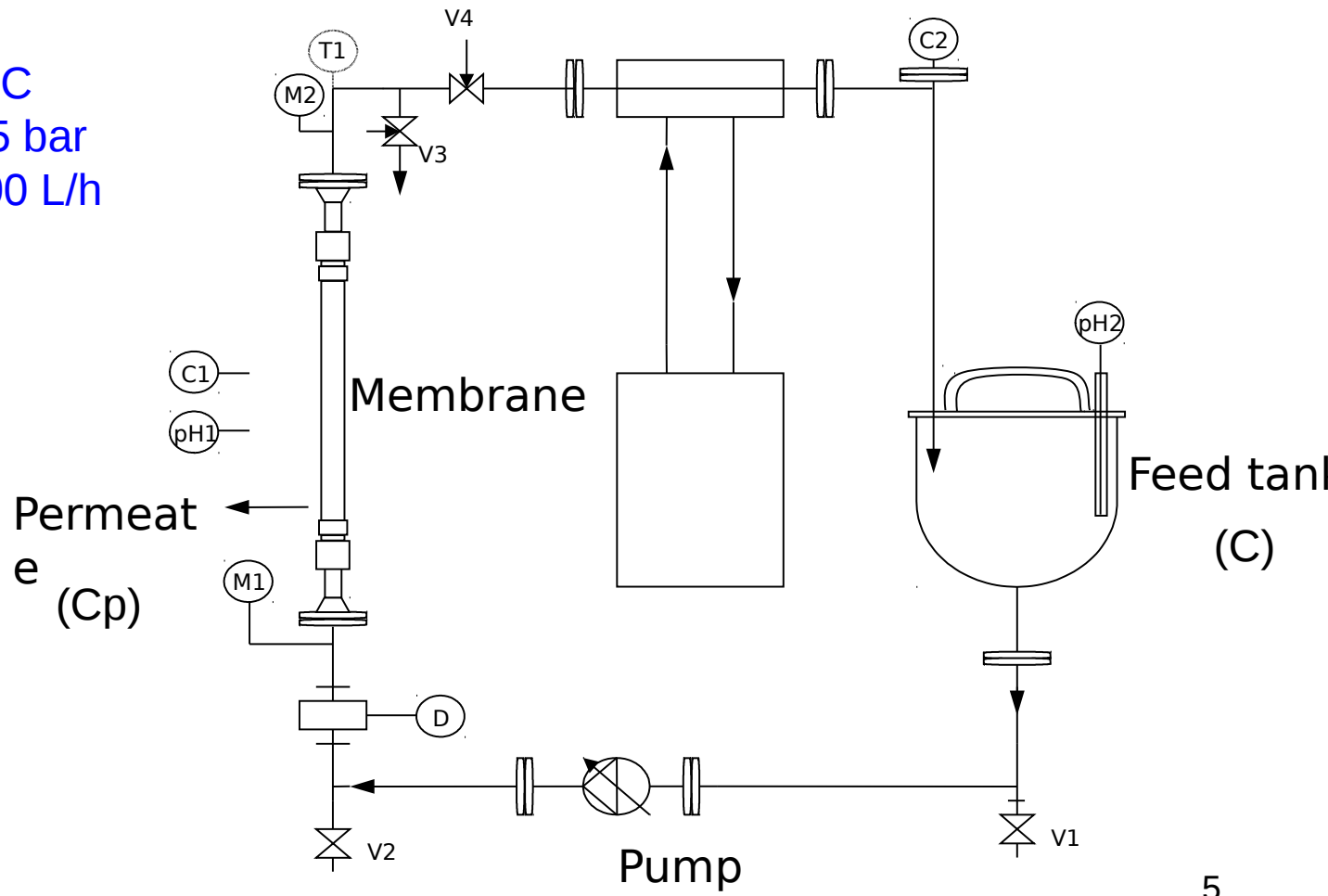
- Gradual decline of membrane performances
- Up to complete stop of the installation

- Requires a cleaning process □ waste of time and money, use of chemicals and energy
- Generation of new wastes (solid and liquid)

Experimental section

Experimental setup

$T = 25^{\circ}\text{C}$
 $\Delta P \leq 15 \text{ bar}$
 $Q = 700 \text{ L/h}$



Experimental section

Experimental tests:

Filtration of protein-water solutions

□ rejection rate vs. applied pressure

$$R = \frac{C - C_p}{C}$$

After each test

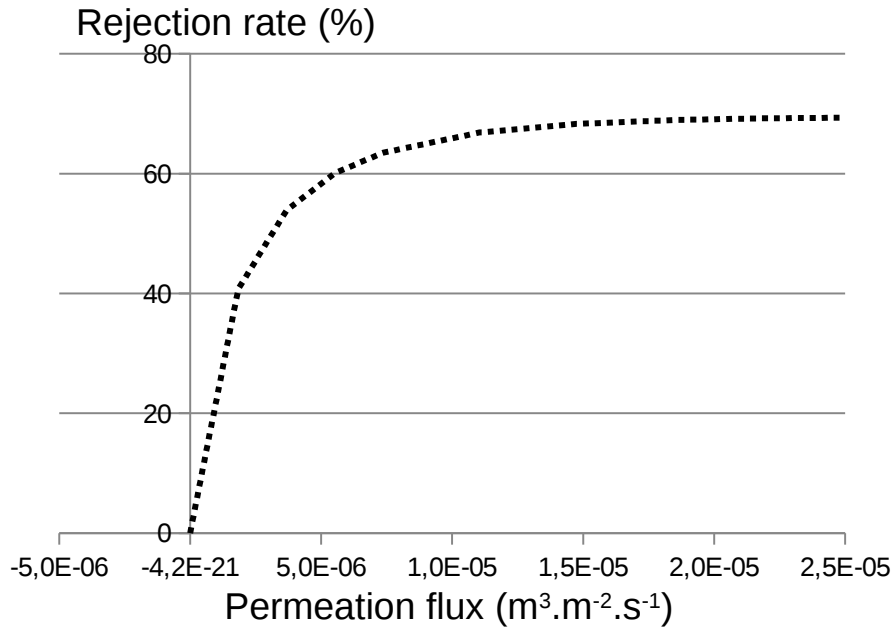
1 - Filtration of pure water □ estimation of hydraulic performances □ permeation flux (J_w) vs. applied pressure (ΔP)

2- Filtration of a neutral solute (Vitamin B12) □ rejection rate vs. applied pressure

□ Nernst-Planck model for neutral molecule □ r_p

Experimental results

Filtration tests of a neutral solute (Vitamin B12)



Nernst-Planck approach for neutral solute

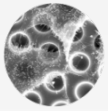
$J(i) = f(\text{convection, diffusion, steric effect})$

solute mass balance + equality chemical potentials

$$\square R = f(r_s, \Delta P, \mu, r_p)$$

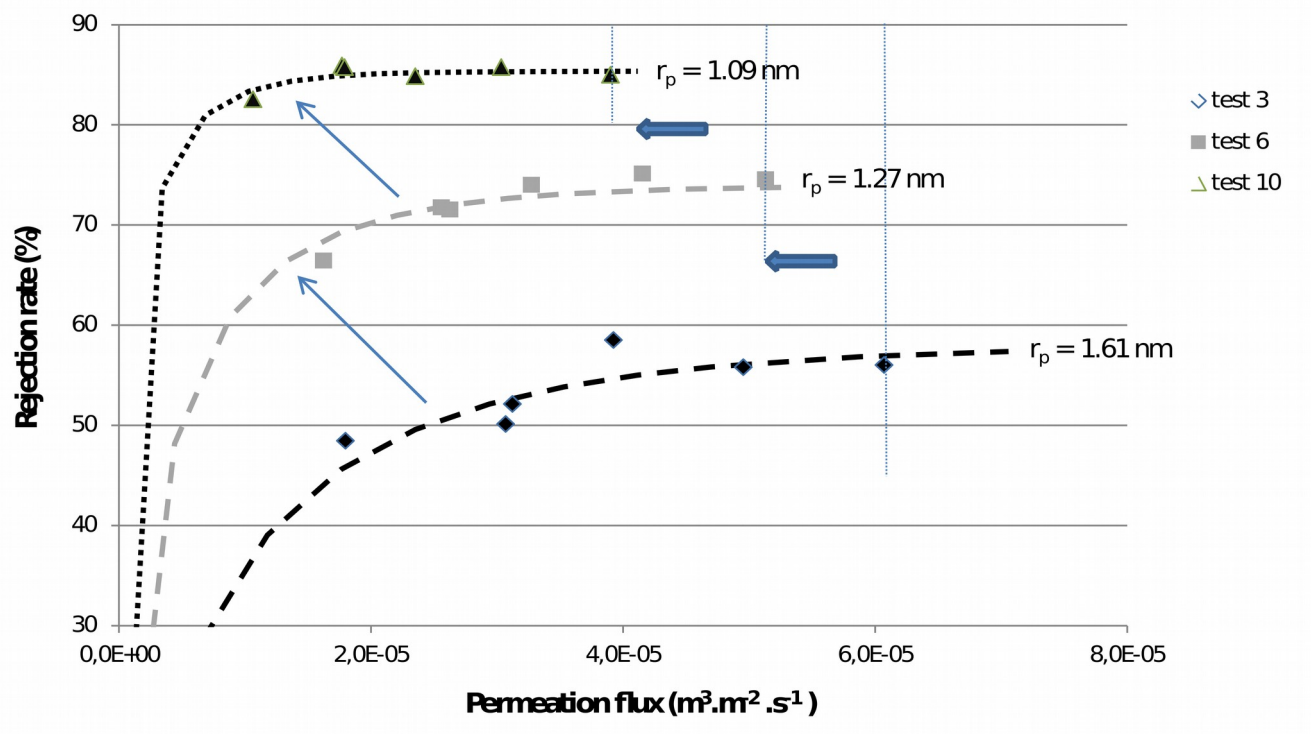
Approximating $R_{exp} = f(\Delta P)$ by the equation $\square r_p$





Experimental section

3 filtration tests of lysozyme



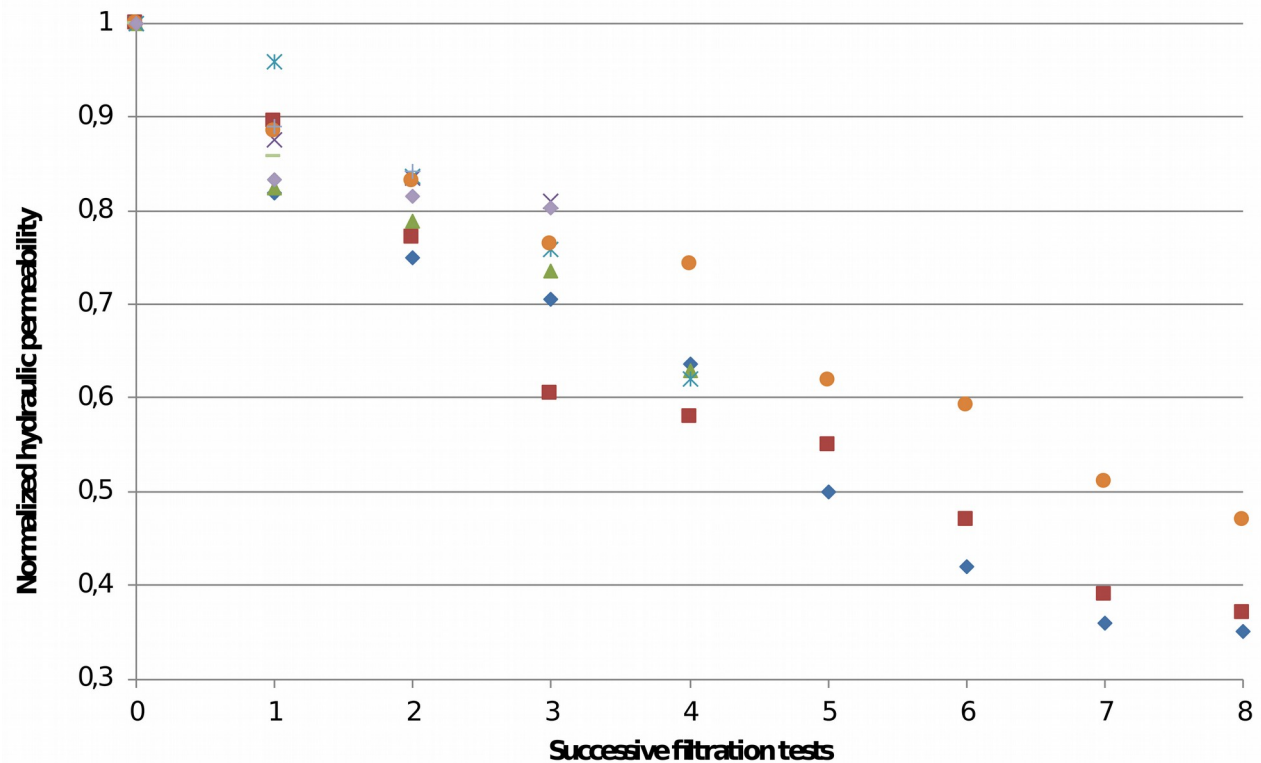
Increase of selectivity

$R(\text{VB12}) = 54 \text{ to } 85 \%$ and $R(\text{lysozyme}) = 65 \text{ to } 100\%$

Hydraulic performances decline ($> 30\%$)

Experimental results

Tests with 7 fresh membranes; filtration of Lysozyme and L-tyrosine (green - orange)



After 7-8 successive tests □ hydraulic performances (60-70 % decline)

Experimental results

Adsorption reversible / irreversible

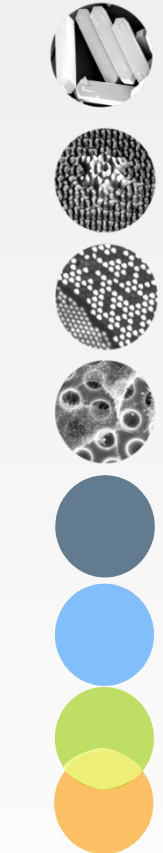
Regeneration tests

hydrothermal treatment (100°C - 5 days)

Use of surfactant

Acid / base cleaning at room T

Acid / base cleaning at high temperature (> 80 °C)



Experimental results

Adsorption of proteins or amino-acids in membrane pores

- adsorption almost irreversible
- rapid performance downgrading
- ideal breeding ground for waterborne bacteria

Main issues / challenges

- understand adsorption phenomena in the pore
- limit pore clogging / or reversibility
- surface modification to reduce protein / surface affinity

Experimental results

Selectivity increases & hydraulic permeability decreases:

Average pore radius decreases

□ little or no adsorption at the membrane surface

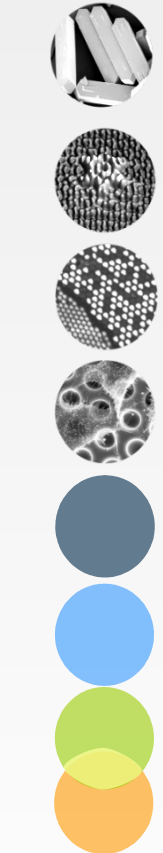
□ size pore distribution

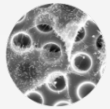
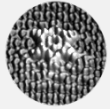
□ adsorption in the largest pore

□ no adsorption in the smallest size pores

Continuous vs. Discontinuous operation

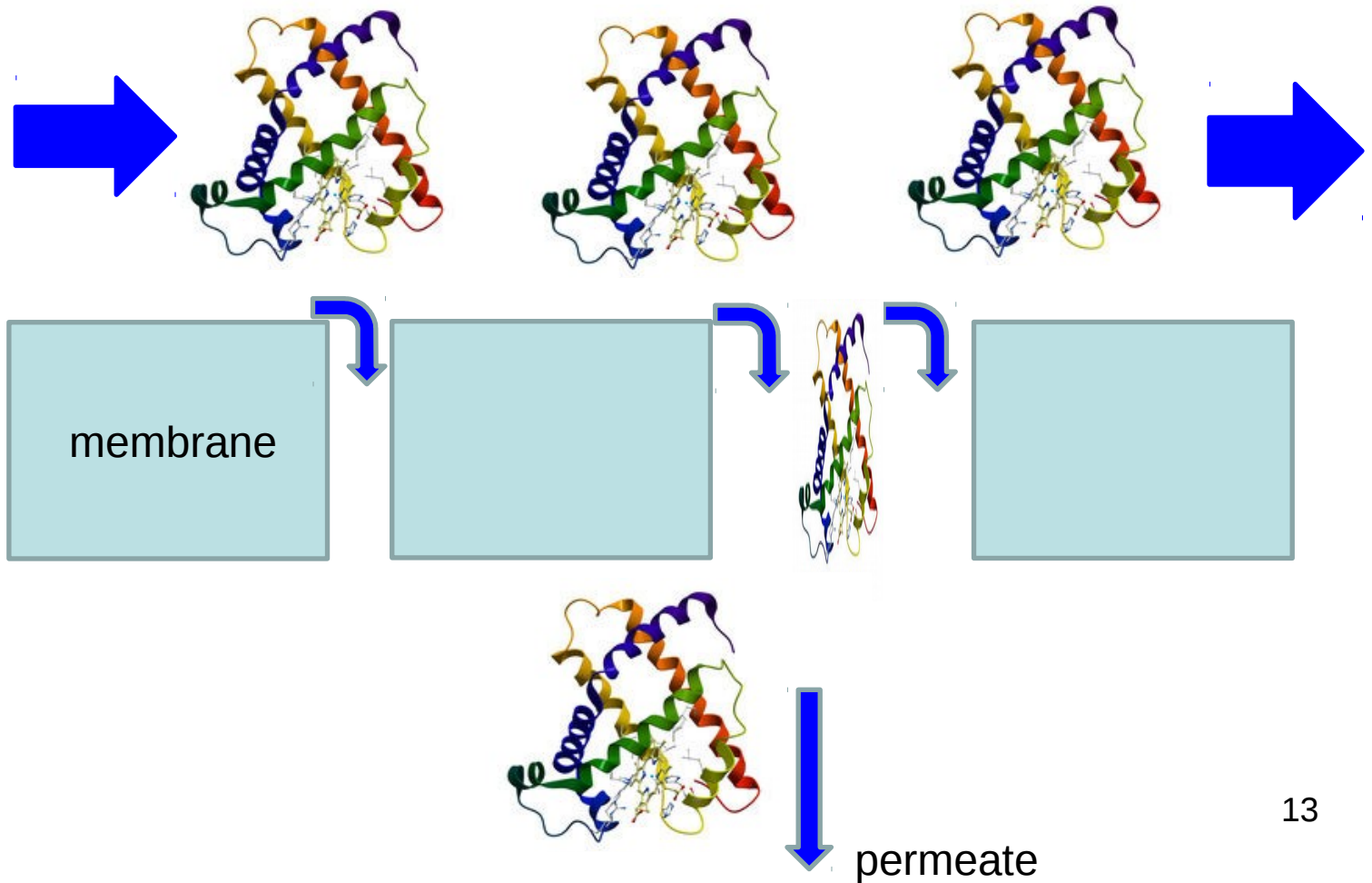
→ filtration during 24 hours or 2 hours = same results

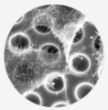
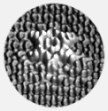




Experimental results

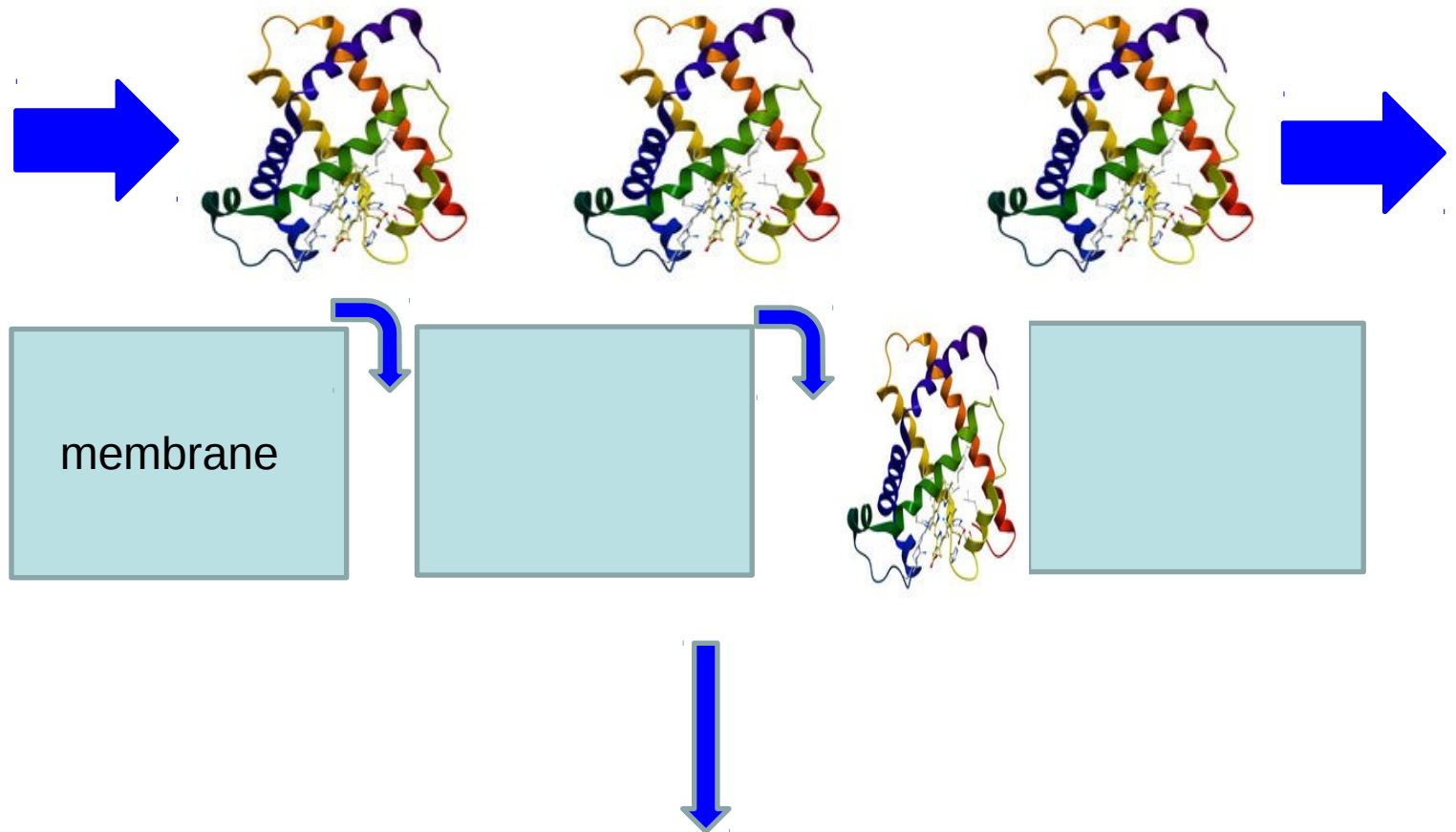
Adsorption phenomena (unsteady) 1st filtration test

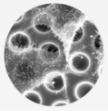




Experimental results

After relaxation





Conclusion

Filtration tests of proteinic solutions

- membrane selectivity □
- hydraulic performances □

Due to

- adsorption in the largest pores
- no adsorption in the smallest and at the surface

- modification of protein conformation

And consequently

- requires treatment for membrane regeneration



Thank you for your
attention